

# BOR4STORE: Fast, Reliable and Cost effective Boron Hydride based high capacity Solid state Hydrogen Storage Materials



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## General

Project funded by the European "Fuel Cells and Hydrogen Joint Undertaking"

→ Total Budget 4.07 Mio.€, total funding 2.3 Mio. €

→ Runtime April 2012 to March 2015

→ 3 Industry partners, 6 Research Institutes

→ Integrated approach for development and testing of novel, optimised and cost-efficient boron hydride based hydrogen storage materials with superior performance (materials capacity more than 8 wt.% and 80 kg H<sub>2</sub>/m<sup>3</sup>) for specific fuel cell applications.

## Approach

- (a) new methods for the synthesis and modification of stable and unstable boron hydrides, as well as their combinations resulting in Reactive Hydride Composites and eutectic mixtures,
- (b) systematic and rationalised investigation of the effect of special catalysts and additives, and
- (c) adaptation of scaffolding concepts.

BOR4STORE aspires to tackle the S&T challenges that still hinder the practical use of the extremely attractive boron hydrides. The technical objectives of the project reflect an innovative and carefully designed strategy involving

## Selected results after 1<sup>st</sup> year

### Eutectically Melting Composites: effect of scaffolding on Li- and Ca-Borohydride mixture

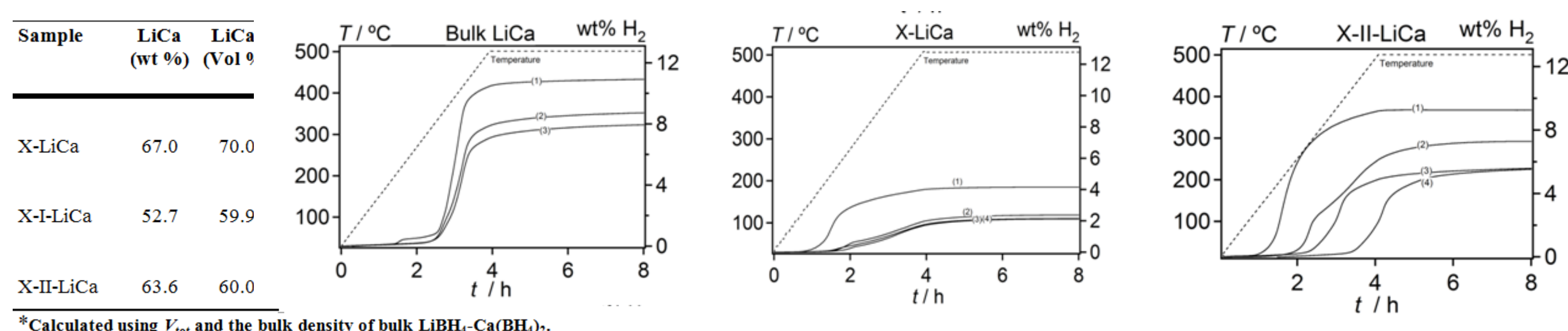


EMC melts around 200 °C

Melt infiltration RT-210 °C, p(H<sub>2</sub>) = 110-150 bar

Dehydrogenation: RT - 500 °C, p(H<sub>2</sub>) = 10<sup>-2</sup> bar

Rehydrogenation: RT - 400 °C, 10 h, p(H<sub>2</sub>) = 180 bar

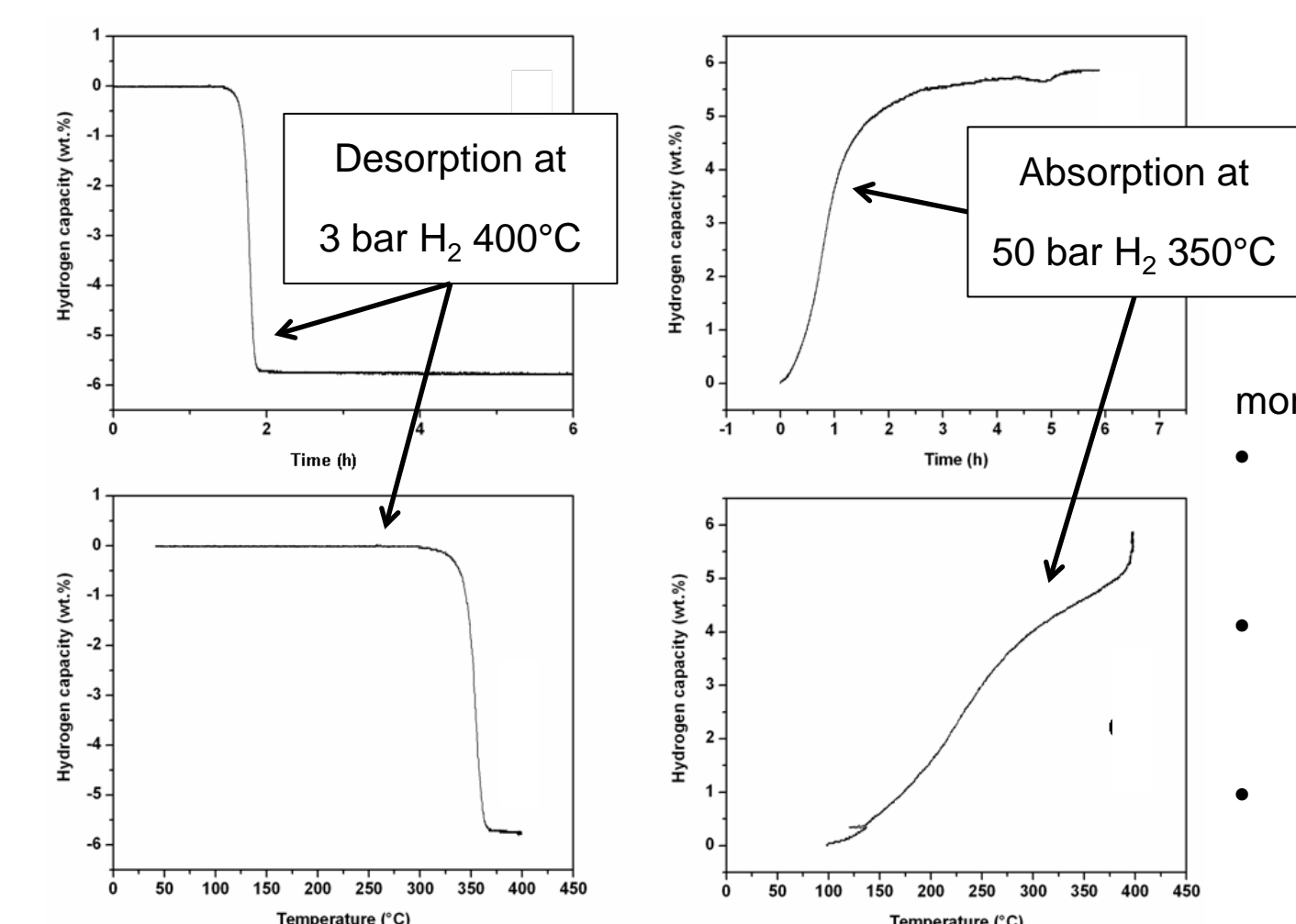


more: 1.18 Payam Javadian, „Scandium Functionalized Carbon Aerogels for Hydrogen Storage“

### Decrease of materials cost ⇒ Recycling of waste materials

- A) As received from turning
- B) Milled
- C) H<sub>2</sub> charged

Mg + 10 wt.% Gd ⇒ MgH<sub>2</sub> + GdH<sub>2</sub> mixture

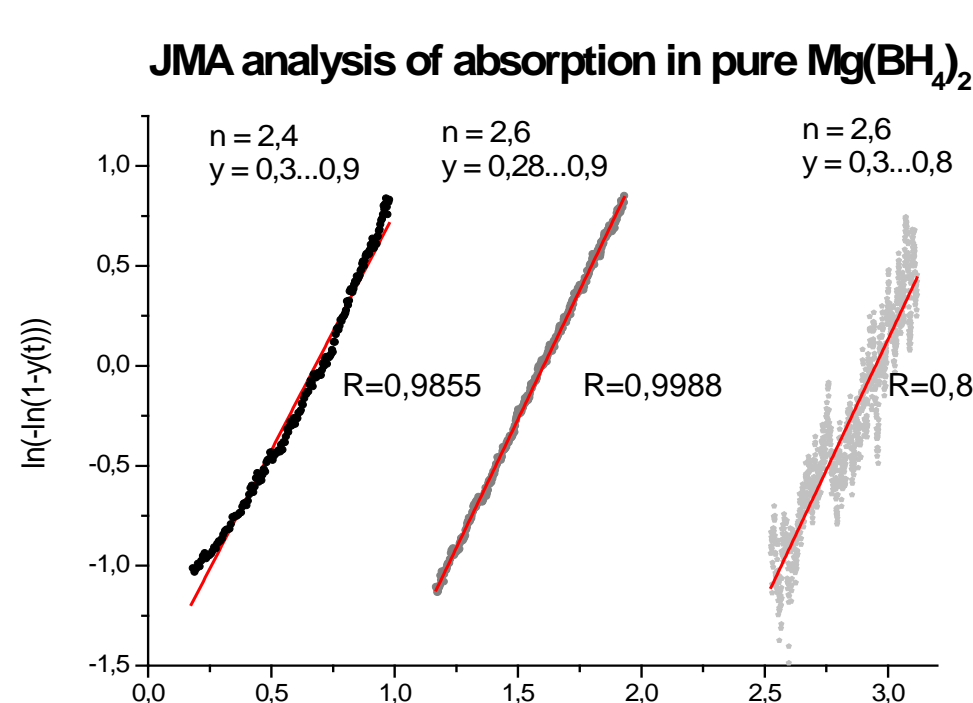
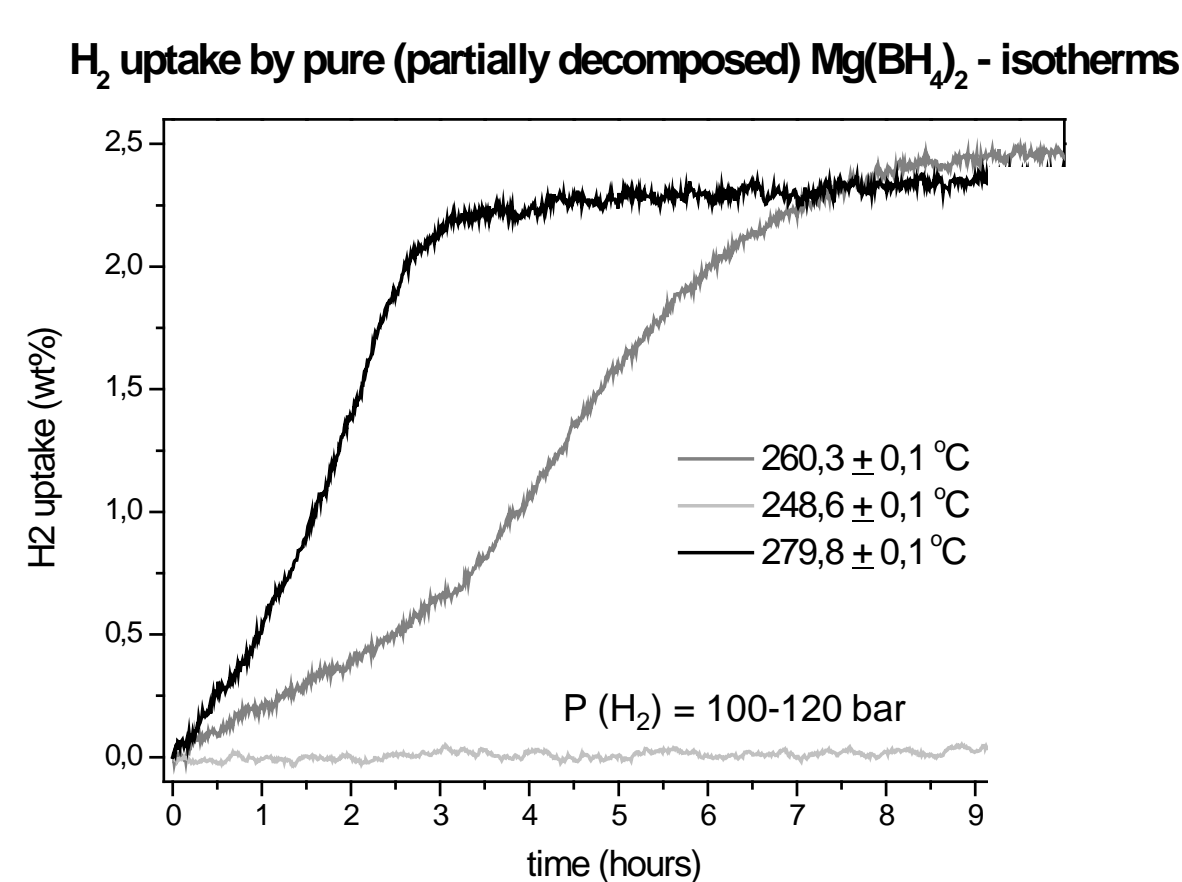


- 2.38 C. Pistidda, „Production of H<sub>2</sub> storage materials from waste Mg alloys“
- 1.46 N. Bergemann, „NaAlH<sub>4</sub> production from waste aluminum by reactive ball milling“
- 1.32 M. Dornheim, „Development and Characterization of Novel Materials for Hydrogen Storage“

### Mg(BH<sub>4</sub>)<sub>2</sub>: rate limiting steps in reabsorption



H<sub>2</sub> absorption in partially decomposed Mg(BH<sub>4</sub>)<sub>2</sub>

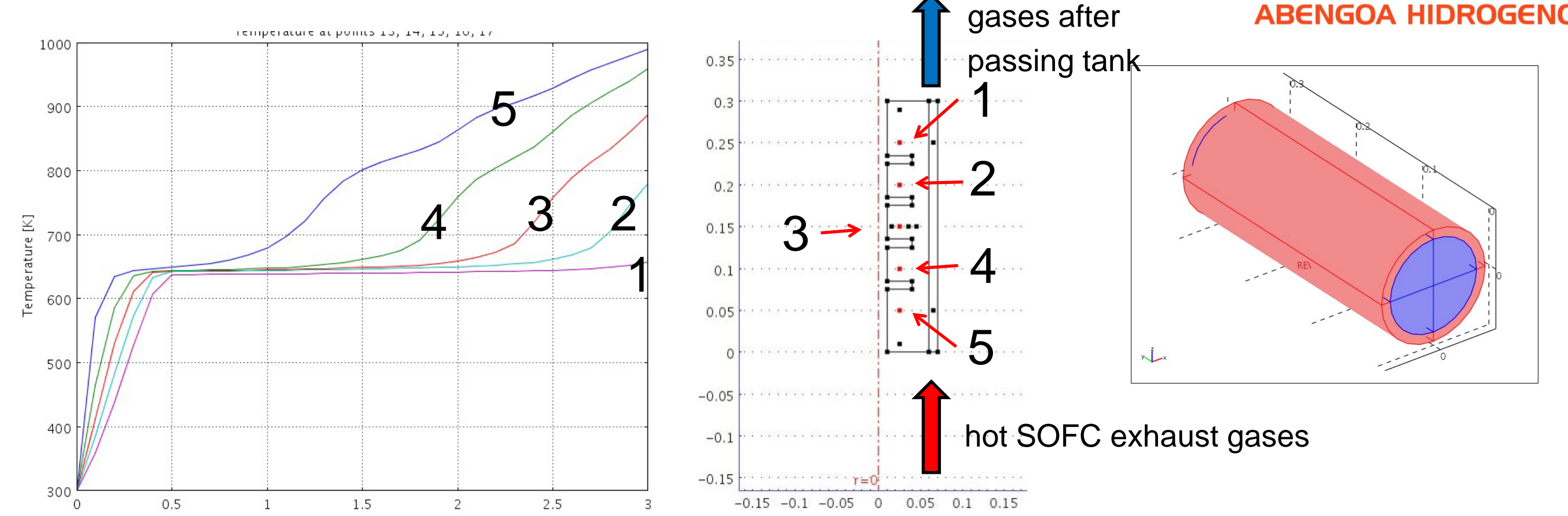


Nucleation at a constant rate and diffusion-controlled growth (n = 5/2)

more: 2.21 Olena Zavorotynska, „Effect of additives on hydrogen sorption in Mg(BH<sub>4</sub>)<sub>2</sub>“

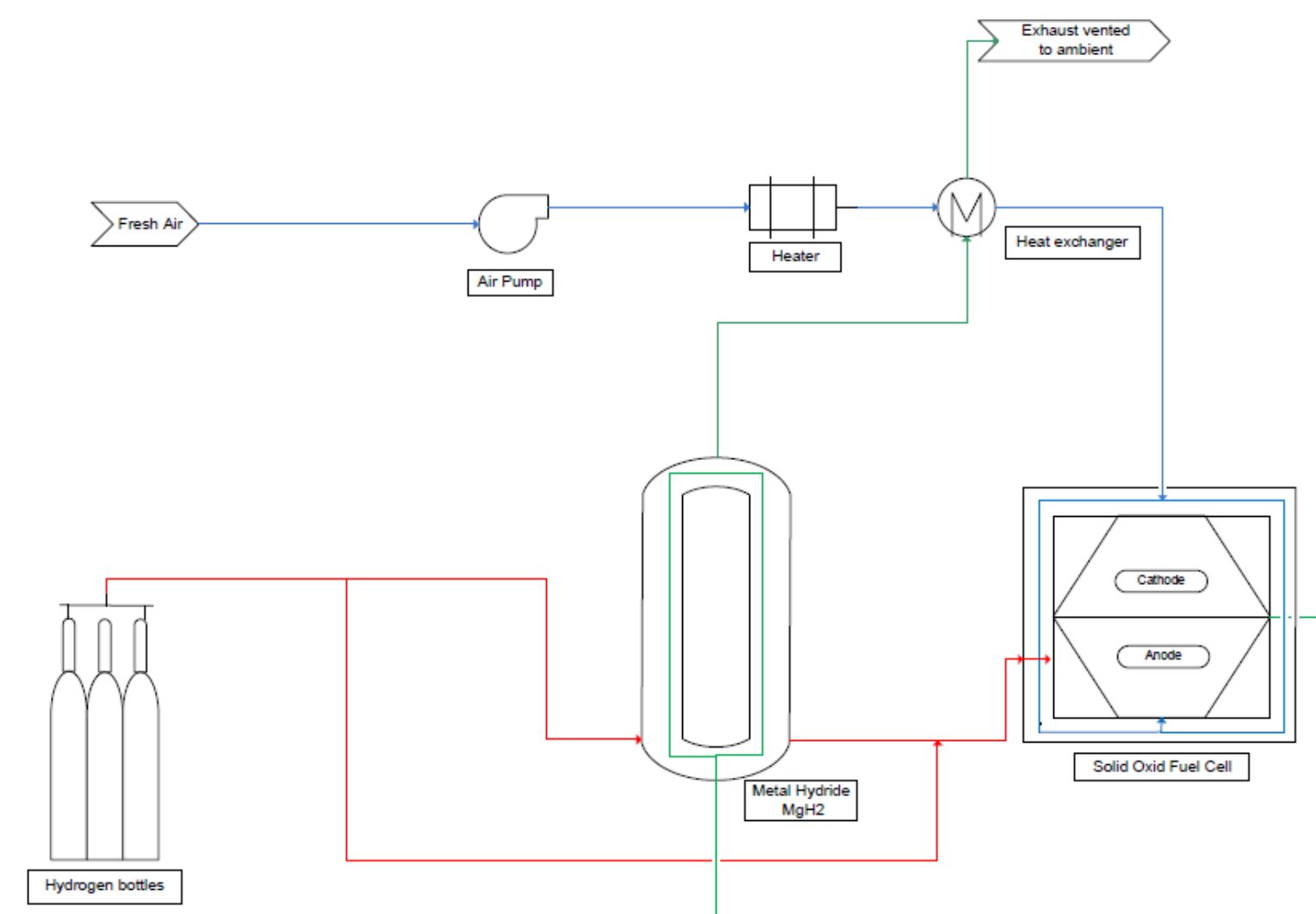
### Simulation of combined SOFC – solid state hydrogen storage (SSHS) system

including thermo-chemical model of the MgH<sub>2</sub> storage material. External heating by exhaust gases, internal resistive heating plus interior fins



parallel flow of exhaust gases leads to inhomogenous unloading  
internal heating necessary to achieve requested hydrogen flow

### First design draft of integrated SOFC – tank system



- 1.7. J. M. Bellosta von Colbe, „Scale-up of Solid-State Hydrogen Storage Tanks“
- 1.35 J. Jepsen, „Design and evaluation of a LiBH<sub>4</sub> - MgH<sub>2</sub> storage system“

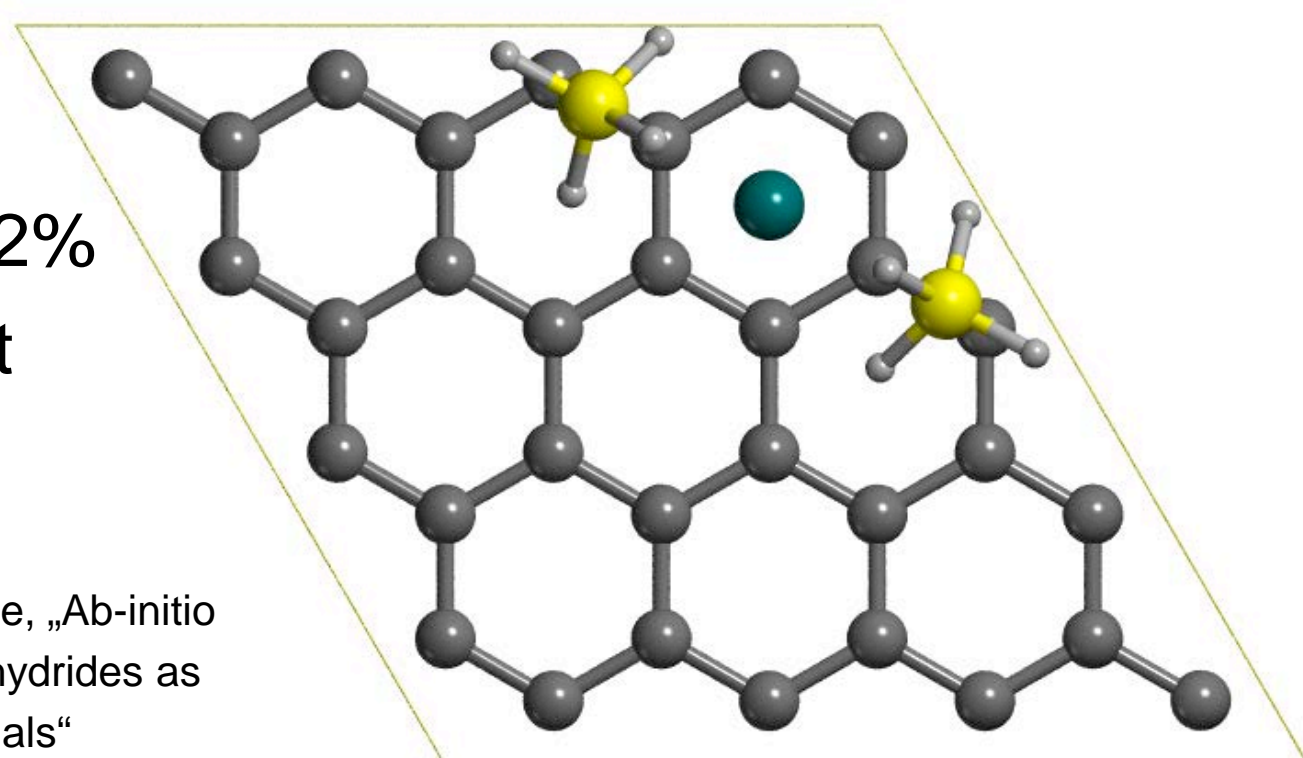
### Modelling

#### Confinement

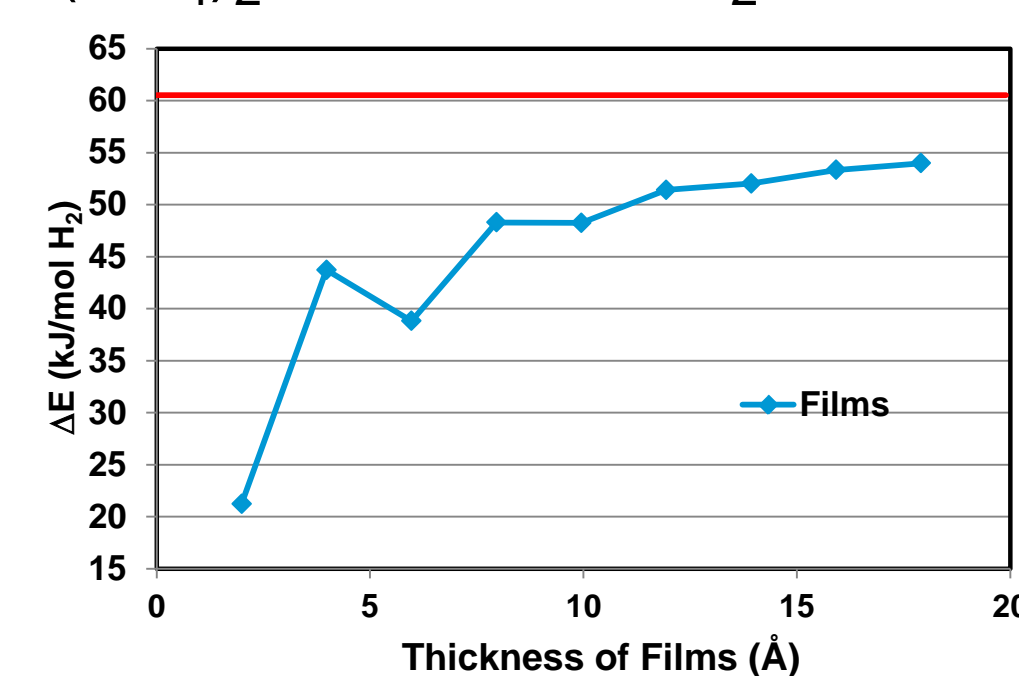
nanoconfinement of Ca(BH<sub>4</sub>)<sub>2</sub> in carbon scaffold  
⇒ simple model system containing a graphite monolayer and the smallest cluster of Ca(BH<sub>4</sub>)<sub>2</sub>

B—H<sub>slab</sub> + 1.2%  
with respect  
to the bulk

more: 2.1. Elisa Albanese, „Ab-initio modelling of metal borohydrides as hydrogen storage materials“



#### Nanostructure



The nano-structuring of Ca(BH<sub>4</sub>)<sub>2</sub> in thin films decreases the decomposition enthalpy

## Summary and further targets

Novel solid state hydrogen storage prototype system based on boron hydrides

→ System capacity > 40 kg H<sub>2</sub>/m<sup>3</sup>, > 4 wt.% with priority on volumetric cap.

⇒ > 80 kg H<sub>2</sub>/m<sup>3</sup>, > 8 wt.% on materials level

→ Materials reaction enthalpies and kinetics of hydrogen loading and discharge suitable for typical load cycles of SOFC in net independent power supply

→ Cycling stability >98% of retained capacity over at least 500 loading-unloading cycles

Cost effective production route of the hydrogen storage material

→ Use of low purity raw materials

→ Demonstration of potential for scale-up of production and system cost of 500 €/kg of stored H<sub>2</sub>

Laboratory prototype of SOFC integrated with hydrogen storage system

→ Model for a continuous power supply

→ Power in the range 0.1 – 1 kW ⇒ tank system ca. 100 - 1000 NI

Compared to compressed gas storage and other fuel cell technologies, respectively

→ Improved storage capacity

→ Improved overall energy efficiency

→ Decreased total cost of ownership

→ Indicator of allowable hydrogen purity for stable storage properties

Demonstration of

Techno-economical readiness of solid state hydrogen storage technology

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Financial support by the European Fuel Cells and Hydrogen Joint Undertaking (http://www.fch-ju.eu) under collaborative project "BOR4STORE" (Grant agreement no.: N° 303428) is thankfully acknowledged.